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### Feasibility of a HMO-process in drinking water treatment technology for removing natural radioactivity and avoiding generation of NORM

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#### **INTRODUCTION**

- *Ca.* 18 % of Estonian inhabitants (230 000 peoole) consume drinking water where indcative dose (ID) exceeds 0.10 mSv/yr \* (Forte *et al.*, 2010)
  \* Parametric value given by Directive 2013/51/Euratom.
- High ID caused by <sup>226</sup>Ra and <sup>228</sup>Ra in Cambrian-Vendian groundwater.
- Groundwater treatment by filtration results in **NORM accumulation**.
- More than 300 tonnes of filter media with <sup>226</sup>Ra, <sup>228</sup>Ra, <sup>228</sup>Th ≥ 1 kBq/kg

(Leier et al., 2018)

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Average activity concentrations <sup>226</sup>Ra – 7.6 kBq/kg, <sup>228</sup>Ra – 8.0 kBq/kg <sup>228</sup>Th – 5.6 kBq/kg

- No national NORM management strategy.
- Potential for uncontrolled NORM waste disposal.

### **AIMS AND OBJECTIVES**



- Ideal solution: a NORM-free water treatment technology.
- LIFE Alchemia "Towards a smart & integral treatment of natural radioactivity in water provision services" (2017-2020) <u>https://www.lifealchemia.eu/en/</u>
  - Objective: to demonstrate the technical and economic feasibility water treatment technologies to remove radioactivity from water and to minimize the generation of NORM.
  - 3 pilot plants in Spain and 1 pilot plant in Estonia.
  - Technology chosen in Estonia the HMO process.



#### THE HYDROUS MANGANESE OXIDES PROCESS **(IDENTIFY OF TARTU**) UNIVERSITY OF TARTU

- For Fe, Mn, and Ra removal
- 3-stage process: aeration  $\rightarrow$  HMO oxidation  $\rightarrow$  filtration
- **HMO suspension**: 2KMnO<sub>4</sub> + 3(MnSO<sub>4</sub>·H<sub>2</sub>O) → 5MnO<sub>2</sub>↓ + K<sub>2</sub>SO<sub>4</sub> + 2H<sub>2</sub>SO<sub>4</sub> + H<sub>2</sub>O
- Removal mechanisms:
  - $4Fe(HCO_3)_2 + O_2 + 2H_2O \leftrightarrow 4Fe(OH)_3\downarrow + 8CO_2$
  - $2Mn(HCO_3)_2 + O_2 + 2H_2O \leftrightarrow 2Mn(OH)_4 + 4CO_2$
  - Ra adsorption to and co-precipitation with Fe and Mn particulates and MnO<sub>2</sub>

#### THE HYDROUS MANGANESE OXIDES PROCESS **(DECESS)** UNIVERSITY OF TARTU



Best removal rates in **lab-scale experiments**:

- Mn 81%
- Fe 90%
- Ra 86%

#### **Schemes tested on the pilot plant:**

- a) Aeration  $\rightarrow$  HMO oxidation  $\rightarrow$  Filtration [gravel-sand-anthracite], periodic dosing
- b) Aeration  $\rightarrow$  HMO oxidation  $\rightarrow$  Filtration [gravel-sand-anthracite-HMO precipitates], periodic dosing
- c) Aeration  $\rightarrow$  HMO oxidation  $\rightarrow$  Filtration [gravel-sand-anthracite], continuos dosing

### **THE PILOT PLANT**

- Situated in Viimsi, Estonia in a water treatment facility operated by Viimsi Water Ltd.
- Target values in treated water

Fe	<b>Mn</b>	<b>ID</b>	
(μg/L)	(μg/L)	(mSv/yr)	
≤ 200	≤ 50	≤ 0.10	

 Raw water from Cambrian-Vendian aquifer

Fe	Mn	<sup>226</sup> Ra	<sup>228</sup> Ra	ID
(µg/L)	(µg/L)	(mBq/L)	(mBq/L)	(mSv/yr)
40-420	50-230	240-620	360-840	0.23-0.55



#### **RESULTS – WATER**



Removal efficiencies (%) for Fe, Mn, and Ra isotopes

Green – complient with drinking water quality parameters.

Red – target value exceeded.

	Scheme	Fe	Mn	<sup>228</sup> Ra	U( <sup>228</sup> Ra), k=2
a)	Aeration → HMO → Filtration [gravel–sand –anthracite], periodic dosing	79	79	49	10
		73	89	37	14
		85	83	63	7
		88	90	64	10
b)	Aeration → HMO → Filtration [gravel–sand –anthracite–HMO	83	47	45	13
		88	66	71	9
	periodic dosing	52	95	64	11
c)	Aeration → HMO → Filtration [gravel–sand –anthracite], continuos dosing	92	92	79	4
		87	87	74	7

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#### **RESULTS – BACKWASH WATER**



Activity concentration of <sup>226</sup>Ra and <sup>228</sup>Ra in filter backwash water

	Scheme	<sup>226</sup> Ra (Bq/L)	U( <sup>226</sup> Ra), k=2 (Bq/L)	<sup>228</sup> Ra (Bq/L)	U( <sup>228</sup> Ra), k=2 (Bq/L)
a)	Aeration → HMO → Filtration [gravel–sand –anthracite], periodic dosing	5.48	0.14	7.89	0.33
		5.87	0.19	8.08	0.38
		3.12	0.10	4.23	0.21
b)	Aeration → HMO → Filtration [gravel–sand –anthracite–HMO precipitates], periodic dosing	3.24	0.11	4.32	1.65
		3.83	0.13	6.23	0.31
		2.43	0.08	3.23	0.16

#### **RESULTS – FILTER MATERIAL**



Accumulation rate of <sup>226</sup>Ra, <sup>228</sup>Ra, and <sup>228</sup>Th, in filter material

		<sup>226</sup> Ra	<sup>228</sup> Ra	<sup>228</sup> Th
	Filter material	(Bq/kg yr⁻¹)	(Bq/kg yr⁻¹)	(Bq/kg yr <sup>-1</sup> )
Pilot plant	Sand	160	220	28
	Anthracite	1300	1500	130
Current	Filtersorb®FMH	1400	1860	
Viimsi Water Ltd. (Hill <i>et al.</i> , 2018)	Zeolite	3400	4760	

#### **TO BE CONTINUED...**



#### Experiments will be continued:

- Continuos dosing
- Scheme with 2-stage filtration
  Aeration → HMO oxidation →
  - $\rightarrow$  Filtration I [gravel-sand-anthracite]  $\rightarrow$
  - → Filtration II [gravel-zeolite]





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## Thank you!







#### **References**

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**Forte**, M., Bagnato, L., Caldognetto, E., Risica, S., Trotti, F., Rusconi, R., 2010. Radium isotopes in Estonian groundwater: measurements, analytical correlations, population dose and a proposal for a monitoring stradegy. *Journal of Radiation Protection*, 30, pp. 761-780.

**Hill**, L., Suursoo, S., Kiisk, M., Jantsikene, A., Nilb, N., Munter, R., Realo, E., Koch, R., Putk, K., Leier, M., Vaasma, T., Isakar, K., 2018. Long-term monitoring of a water treatment technology designed for radium removal – removal efficiencies and NORM formation. *Journal of Radiological Protection*, 38, pp. 1–24.

**Leier**, M., Kiisk, M., Suursoo, S., Vaasma, T., Putk, K., 2018. Formation of radioactive waste in Estonian water treatment plants. *Journal of Radiological Protection*, 39, pp. 1–10.